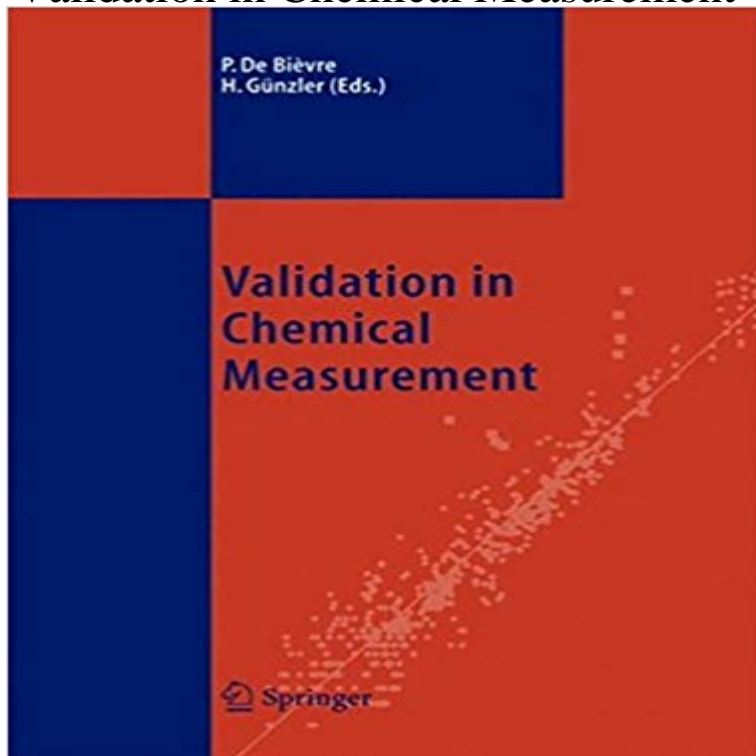


# Validation in Chemical Measurement



Validation of measurement methods has been associated with measurement uncertainty? The answer must be: no, not in chemistry.

It is mostly based on the examination of a measurement procedure for its characteristics such as automatic validity, precision, accuracy, selectivity, sensitivity, repeatability, and the skills, the role and the responsibility of the analyst.

productivity, detection limit, quantification limit and more. ISO 9000:2000, item 3.8.5 defines validation as confirmation. When focussing on quality comparability and reliability by examination and provision of objective evidence in chemical measurement, the fields of interest to this Journal - that the requirements for an intended use are fulfilled. The result, one stumbles into various interpretations of the term. The revised edition of the VIM (VIM3), is likely to define validation. It is one more example of a term which is used in different ways. The definition of the concept validation to be confirmed sometimes very consistently, sometimes very loosely or in- through examination of a given item and provision of - deed ambiguously. Since the term is very common in the field of objective evidence that it fulfills the requirements for a stated chemical community, it is important that its meaning be intended use. And clear. Turning to the 2nd edition of the International Vocabulary of Basic and General terms in Metrology (VIM)

a formal decision that a given measurement method automatically (1993), surprisingly we do not find a definition. Webster's dictionary gives them valid i. e. reliable results. One wonders. The Dictionary of the English language (1992) tells us that validation - what this has to do with stated intended use. Reliability

validation is making or being made valid. Obviously validity clearly is a property of a measurement result.

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